

## COMMUNICATIONS TO THE EDITOR

**Phomactin E, F, and G: New Phomactin-group PAF Antagonists from a Marine Fungus *Phoma* sp.**

Sir:

Platelet activating factor (PAF) is a naturally occurring etherphospholipid that is a mediator of anaphylaxis released by a number of stimulated cells, such as basophils, neutrophils, platelets, and macrophages, and it causes degranulation of polymorphonuclear leukocytes, smooth muscle contraction, vascular permeability, and hypotension.<sup>1)</sup> Intensive efforts to find drugs that attenuate the effects of PAF have resulted in the discovery of a number of specific PAF antagonists, some of which are being tested for their clinical effectiveness.<sup>2,3)</sup>

In the course of finding PAF antagonists from metabolites of a marine fungus *Phoma* sp., we have discovered novel PAF antagonists: phomactin A, B, B<sub>1</sub>, B<sub>2</sub>, D.<sup>4,5)</sup> CHU *et al.* also reported several PAF antagonists with the same skeletal framework as phomactins.<sup>6,7)</sup> Recently we found three related compounds; phomactin E (1), F (2), and G (3). In this paper, we report the structures and PAF antagonistic activities of 1, 2, and 3.

Fermentation was carried out at 23°C with agitation at 80 rpm for 12 days in two 600-liter tanks, each containing 300 liters of a medium consisting of sucrose 2.0%, K<sub>2</sub>HPO<sub>4</sub> 0.5%, peptone 1.0%, potato 1.0%, and CB442 0.02% (pH 8.5).

Culture filtrate (600 liters) was extracted with EtOAc (600 liters). Assay-directed purification of the EtOAc extract by silica gel and reversed-phase chromatography

gave phomactin E (1) (940.0 mg), F (2) (25.0 mg) and G (3) (170.0 mg). The <sup>1</sup>H and <sup>13</sup>C NMR (Tables 1 and 2) and other spectral data (Table 3) showed that these compounds had the same skeletal framework as phomactin B (4).

The molecular formula of phomactin E (C<sub>20</sub>H<sub>30</sub>O<sub>3</sub>), determined by high-resolution mass spectrum (HREI-MS, *m/z* 318.21684;  $\Delta$  = 2.6 mmu), was less by one oxygen than that of 4 (C<sub>20</sub>H<sub>30</sub>O<sub>4</sub>). Comparison of NMR spectra of 4 and 1 showed that one carbon containing a

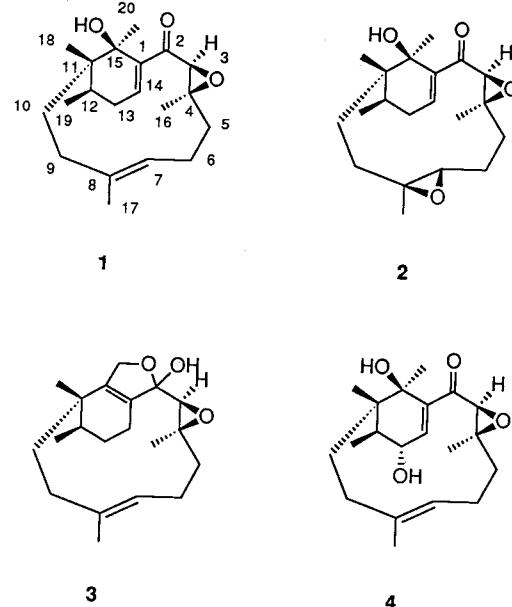


Table 1. <sup>1</sup>H NMR spectra of phomactin E (1), F (2), G (3), and B (4) (CD<sub>3</sub>OD).

Number	(1) ppm (mult., <i>J</i> , Hz)	(2) ppm (mult., <i>J</i> , Hz)	(3) ppm (mult., <i>J</i> , Hz)	(4) ppm (mult., <i>J</i> , Hz)
3	3.91 (s)	3.99 (s)	2.85 (s)	3.81 (s)
5	1.24 (m)	1.38 (m)	1.92 (m)	1.24 (m)
	2.17 (m)	2.23 (m)	1.72 (m)	2.15 (m)
6	2.46 (m)	2.21 (m)	1.89 (m)	2.38 (m)
	2.12 (m)	1.48 (m)	1.79 (m)	2.10 (m)
7	5.37 (br t, 7.2)	2.73 (dd, 10.5, 3.7)	5.07 (br d, 10.0)	5.31 (br t, 7.3)
9	2.20 (m)	1.59 (m)	2.45 (m)	2.16 (m)
	2.20 (m)	1.59 (m)	1.94 (m)	2.16 (m)
10	1.35 (m)	1.99 (m)	1.67 (m)	1.39 (ddd, 15.7, 7.6, 4.6)
	2.17 (m)	1.03 (m)	1.53 (m)	2.02 (ddd, 15.7, 8.0, 4.9)
12	1.61 (dq, 8.5, 7.5)	1.74 (dq, 9.3, 7.3)	2.13 (ddq, 12.0, 2.5, 7.3)	1.63 (dq, 3.2, 7.5)
13	2.78 (ddd, 20.6, 8.5, 3.5)	2.49 (ddd, 19.4, 9.3, 6.4)	1.64 (dq, 12.5, 12.0)	4.12 (dd, 3.2, 2.6)
	2.05 (dd, 20.6, 4.3)	2.10 (dd, 19.4, 2.7)	1.44 (m)	
14	5.86 (dd, 3.5, 4.3)	5.84 (dd, 6.4, 2.7)	2.01 (m)	5.91 (d, 2.6)
			1.85 (m)	
16	1.20 (s)	1.27 (s)	1.40 (s)	1.24 (s)
17	1.63 (s)	1.32 (s)	1.64 (s)	1.60 (s)
18	1.17 (s)	1.15 (s)	0.93 (s)	1.13 (s)
19	1.23 (d, 7.3)	1.18 (d, 7.3)	0.92 (d, 7.3)	1.27 (d, 7.5)
20	1.48 (s)	1.49 (s)	4.51 (ddq, 12.5, 2.0, 1.5)	1.46 (s)
			4.37 (ddd, 12.5, 4.1, 1.3)	

Table 2.  $^{13}\text{C}$  NMR spectra of phomactin E (1), F (2), G (3), and B (4).

Number	1 ppm (mult.)	2 ppm (mult.)	3 ppm (mult.)	4 ppm (mult.)
1	149.5 (s)	150.6 (s)	134.5 (s)	147.2 (s)
2	203.2 (s)	202.6 (s)	109.7 (s)	200.3 (s)
3	68.8 (d)	68.2 (d)	64.9 (d)	65.9 (d)
4	64.2 (s)	64.4 (s)	61.4 (s)	62.8 (s)
5	39.2 (t)	35.5 (t)	35.1 (t)	37.4 (t)
6	24.4 (t)	25.3 (t)	24.4 (t)	22.7 (t)
7	121.0 (d)	62.5 (d)	129.5 (d)	120.3 (d)
8	137.8 (s)	64.0 (s)	135.2 (s)	136.8 (s)
9	33.9 (t)	38.9 (t)	36.1 (t)	33.6 (t)
10	35.0 (t)	35.6 (t)	33.6 (t)	36.7 (t)
11	44.0 (s)	43.7 (s)	38.6 (s)	41.5 (s)
12	40.9 (d)	34.3 (d)	34.5 (d)	46.3 (d)
13	32.9 (t)	33.4 (t)	29.1 (t)	71.4 (d)
14	134.4 (d)	133.4 (d)	24.1 (t)	135.6 (d)
15	74.2 (s)	76.5 (s)	144.7 (s)	73.3 (s)
16	15.3 (q)	14.8 (q)	21.0 (q)	14.5 (q)
17	17.9 (q)	16.2 (q)	17.0 (q)	16.4 (q)
18	19.0 (q)	21.1 (q)	22.2 (q)	19.7 (q)
19	21.1 (q)	22.0 (q)	14.9 (q)	19.7 (q)
20	26.0 (q)	23.2 (q)	70.5 (t)	23.2 (q)

1, 2, 3 were measured in  $\text{CD}_3\text{OD}$ . 4 was measured in  $\text{DMSO}-d_6$ .

Table 3. Physicochemical properties of phomactin E (1), F (2), and G (3).

	1	2	3
MP (°C)	148~149	199~202	131~132
Molecular formula	$\text{C}_{20}\text{H}_{30}\text{O}_3$	$\text{C}_{20}\text{H}_{30}\text{O}_4$	$\text{C}_{20}\text{H}_{30}\text{O}_3$
HREI-MS	318.21684	334.21322	318.21765
$[\alpha]_D^{25}$ (CHCl <sub>3</sub> )	+178.4	+120.9	+96.9
UV $\lambda_{\text{max}}$ (ε)	238 (3500)	239 (3200)	End

hydroxyl ( $\delta_{\text{C}}$  71.4 (d),  $\delta_{\text{H}}$  4.12 (1H dd,  $J=3.2, 2.6$  Hz)) in 4 was replaced by a methylene carbon (( $\delta_{\text{C}}$  32.9 (t),  $\delta_{\text{H}}$  2.78 (1H, ddd,  $J=20.6, 8.5, 3.5$  Hz), 2.05 (1H, dd,  $J=20.6, 4.3$  Hz)). Furthermore, the long-range coupling was observed from the doublet methyl proton ( $\delta_{\text{H}}$  1.23) to this methylene carbon. 1 was therefore suggested to be a deoxy derivative of 4 at C<sub>13</sub>. In order to verify the proposed structure and establish the overall stereochemistry of 1, X-ray diffraction analysis was performed on a crystal obtained from  $\text{CH}_2\text{Cl}_2$ -hexane (space group  $P2_12_12_1$ ,  $a=20.382(1)$  Å,  $b=13.498(1)$  Å,  $c=13.391(2)$  Å,  $Z=8$ ). The structure was determined by the direct method (MULTAN 78) and successive block diagonal least-squares and Fourier synthesis. Parameters were refined by using anisotropic temperature factors to  $R=0.060$ . The ORTEP of 1 is shown in Fig. 1.

The X-ray analysis revealed a further structural feature in that  $\alpha, \beta$ -unsaturated ketone (C<sub>14</sub>-C<sub>1</sub>-C<sub>2</sub>-O) was nonplanar; the dihedral angle between the carbonyl and the C<sub>1</sub>-C<sub>14</sub> double bond was 119°. This accounted for the

Fig. 1. ORTEP drawing of phomactin E (1).

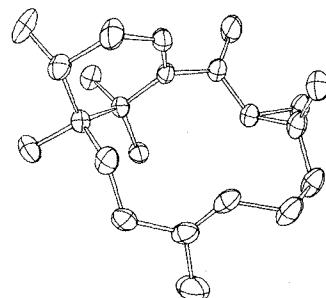
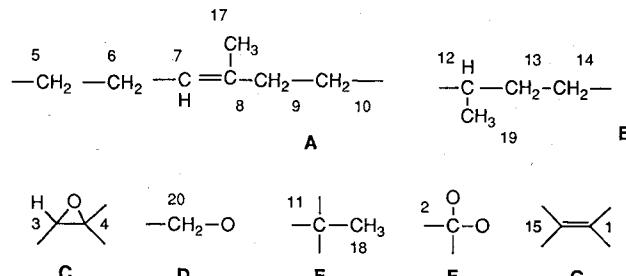


Fig. 2. Partial structures of phomactin G (3).



low  $\epsilon$  (3500) in the UV, and an unusual  $^{13}\text{C}$  NMR assignment at C<sub>1</sub> ( $\delta_{\text{C}}$  149.5) and C<sub>14</sub> ( $\delta_{\text{C}}$  134.4).

Phomactin F (2) has the molecular formula  $\text{C}_{20}\text{H}_{30}\text{O}_4$ , determined by HREI-MS ( $m/z$  334.21322  $\Delta=1.1$  mmu). In the  $^1\text{H}$  and  $^{13}\text{C}$  NMR, 2 differed from 1 only in the replacement of one double bond ( $\delta_{\text{C}}$  121.0 (d), 137.8 (s); 5.37 (1H, br t,  $J=7.2$  Hz) in 1 by a epoxide ( $\delta_{\text{C}}$  64.0 (s), 62.5 (d);  $\delta_{\text{H}}$  2.73 (1H, dd,  $J=10.5, 3.7$  Hz) in 2. The UV spectrum showed the maximum at 239 nm ( $\epsilon$  3200) due to the enone (C<sub>14</sub>-C<sub>1</sub>-C<sub>2</sub>-O). Combining of the above data suggested that 2 had the epoxide at C<sub>7</sub> and C<sub>8</sub>. To confirm the structure, 1 was converted to 2 by MCPBA oxidation.

The MCPBA oxidation of 1 in CHCl<sub>3</sub> gave a single diastereomer in an 87% yield, suggesting that the oxidation occurred at the less hindered side of the double bond. From the ORTEP drawing of 1, the 8-Si plane at C<sub>7</sub>, C<sub>8</sub> was more open and therefore preferred by the attack of MCPBA to give 2. To clarify the conformation of 1 in CHCl<sub>3</sub>, NOE experiments were performed. The irradiation of H<sub>3</sub> resulted in enhancement of H<sub>20</sub>, whereas the irradiation of H<sub>7</sub> resulted in enhancement of H<sub>16</sub>, H<sub>14</sub>, and H<sub>13</sub>. These results showed that 1 had the same conformation in CHCl<sub>3</sub> as in crystal state. Hence the structure of phomactin F, including the stereochemistry at C<sub>7</sub> and C<sub>8</sub>, is proposed as 2.

Phomactin G (3) has the molecular formula  $\text{C}_{20}\text{H}_{30}\text{O}_3$ , from its HREI-MS ( $m/z$  318.21765  $\Delta=1.8$  mmu). The  $^1\text{H}$ ,  $^{13}\text{C}$  NMR, and DQF COSY experiments inferred the partial structure A~G (Fig. 2). Further information regarding the skeletal framework was sought from the COLOC experiment. The cross peak of 3H<sub>18</sub> to C<sub>12</sub>,

$C_{15}$ , and  $C_{10}$  confirmed the linkage **E**, **B**, **G**, and **A**. The linkage **A** and **C** was obtained from the coupling of  $3H_{16}$  with  $C_5$  and  $C_3$ . The couplings of  $H_{20}$  with  $C_{15}$ ,  $C_1$ , and  $C_2$  suggested that **D**, **G**, and **F** constituted a dihydrofuran ring. Insertion of this dihydrofuran ring between  $C_3$  and  $C_{14}$  was based on the coupling of  $H_3$  to  $C_2$  and  $H_{14}$  to  $C_1$ . Based on these data, the structure of phomactin **G** was proposed as **3**. The stereochemistries at  $C_3$ ,  $C_4$ ,  $C_{11}$ , and  $C_{12}$  were assumed to be same as **1** because **1** and **3** may have the same biosynthetic pathway. However the stereochemistry at  $C_2$  is still unknown.

Phomactin **E** (**1**), **F** (**2**), and **G** (**3**) inhibited PAF-induced platelet aggregation with  $IC_{50}$ s of  $2.3\ \mu M$ ,  $3.9\ \mu M$ , and  $3.2\ \mu M$ , respectively, and also inhibited binding of PAF to its receptors with  $IC_{50}$ s of  $5.19\ \mu M$ ,  $35.9\ \mu M$ , and  $0.38\ \mu M$ , respectively. **2** is less active than **1** due to the presence of epoxide at  $C_7$ ,  $C_8$ . These data suggested that the lipophilicity at  $C_7$ ,  $C_8$  had a significant role in receptor binding.

The absolute stereochemistry of **1**, **2**, and **3** is now under investigation.

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(Received May 23, 1995)

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